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O61 ; O641.3

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100 -281X(2011)07-1386-14

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Supramolecular chemistry is defined as “ chemistry beyond the molecule”, bearing on the organized entities of higher complexity that result from the association of two or more chemical species held together by intermolecular forces. Now, the applications of supramolecular chemistry in the realm of radio chemistry have attracted much attention. In this article, the separation of important metal ions by two kinds of supramolecular systems and ionic imprinted polymers (IIP) is discussed, which demonstrates the significant roles of supramolecular systems in the field of radio chemistry. Meanwhile, the first type is supermolecules resulted from the intermolecular association of a host molecule and its substrates, where the host molecules are mainly macrocyclic compounds (such as crown ethers, calixarene, calix crown ethers and so on). The second type is supramolecular assemblies induced by the spontaneous association of a large undefined number of components into a specific phase, including reversed micelles, microemulsions and liquid membranes. IIPs can recognize and separate metal ions selectively, which realizes the recognition function of supramolecular chemistry. The transport function of supramolecular chemistry is also well interpreted in the separation of metal ions by liquid membrane. Besides, the perspectives of supramolecular chemistry in the field of radio chemistry are also presented.

supramolecular chemistry; radio chemistry; separation of metal ions; macrocyclic compounds;

; 2011 1 , ; 2011 3

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(No.20871009) 4

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ioni imprinting

1 Introduction

2 The separation of metal ions by macrocyclic compounds

2.1 Crown ethers

2.2 Calixarenes

2.3 Calix crown ethers

3 The separation of metal ions by supramolecular assemblies

3.1 Reversed microemulsions

3.2 Liquid membranes

4 Ioni imprinted polymers

Perspectives

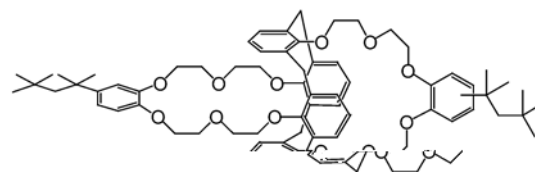
Ů ɔ (supramolecular chemistry) ζ
 ≡ ≡ w ɔ ɔ Ā θ Ů θ
 ' m Ě 4m ' m
 ɔ [1] Ů ɔ ɔ m
 w , A , Ā , Ĥ ɔ
 Ů ɔ (1987 ɔ) ,
 Louis Pasteur Jean-Marie Lehn ε
 “ Ů ɔ ” , “ Ů Ж ” .
 Lehn Ů , Ů ɔ Ů w ≡
 [1] : (1) Ů (supermolecules) , m
 Ů (m ɔ w m) Ů
 ɔ , Ů
 Ě ; (2)
 Ů (supramolecular assemblies) ,
 Ů ' y ()
 Ě m ' Ů , ,
 , , , Ů ě
 ζ Ů : , , 4 ,
 Ů ' 4 .
 ɔ ɔ ɔ Ů , S
 Ů . Cameron Ĥ
 1910 O ɔ (radiochemistry)
 τ , ρ : ɔ w ɛ ɔ y
 ɔ 4 m [2] m
 , ɔ , ɛ

4 m w ɔ (chemistry
 of radioelements) , 4 ɔ .
 (4) ɔ
 , w w Ā w ɔ ;
 ~ (4
 ζ Sr, Cs) ɔ 4 , w w
 Ů , ɔ 4 w ɔ [2] .
 , ɔ ɔ A Δ , ɛ
 Δ “ ” .
 ζ ≡ Ů 4 ○
 ' Ů , Ů ɔ
 ɔ . m Ů
 ζ Ů , , ɔ
 ' . J Ů ζ κ , 4
 . ○ Ů Ů ,
 “ Ĥ ɔ ≡ Ů . Ů
 Ů ' Ů .
 Ů , Ů ' ,
 . Ů ɔ ɔ
 θ Ů .
 1967 Pedersen ' w , ɔ
 ' (, ,)
 . Ĥ Ů O, S, N, ' A
 ' , ' Y “ ”
 “ ” Ě H , , ' w ,
 Ů . m ,
 H , , ɔ ' w Ů
 .
 2.1
 ɔ ' Ů
 ɔ Ā [3—7] , “ ”
 A ' Ů “ ” Binnemans [8]
 “ ” w w
 A UO₂²⁺ ' Ů : UO₂²⁺ , 6
 ' ;
 A UO₂²⁺ H ; A UO₂²⁺
 H ; Ĥ H .
 ɔ Ů ɛ , ' w
 ζ Ů A ' Ů [9,10] . Grüner [9] '
 m “Ω “ ” ,

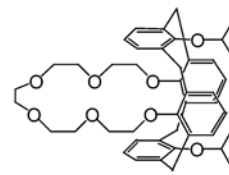
, ɔ' Na⁺ ɣ ; , ɐ
 Cs⁺ ɣ
 Sr²⁺ ɣ ũ ǁ J -18-
 -6(DCH18C6)- ɔ ũ
 Sr²⁺, ɐ ɥ Na⁺, K⁺,
 Cs⁺, Ni²⁺, Fe³⁺, Al³⁺, Cr³⁺, Mo⁶⁺ ɣ .
 DCH18C6 Sr²⁺ ɣ ,
 K⁺ 4 Mo⁶⁺ ' DCH18C6 ũ ɣ ,
 ɐ ɥ "Hω " ɣ . ɥ Sr²⁺ 4 K⁺
 A ε L [11—13] .
 SREX(strontium extra tion) 0.1 M 4,4'-
 (')-J r -18- -6(DtBuCH18C6,
 1) w ɔ 1. M ɐ r (TBP) ɥ
 ISOPAR® L [14,1] ,

J 21- -7(DCH21C7)^[40] ε γ ,
 (PFOA⁻) ε κ ' w ů
 Sr²⁺、UO₂²⁺、Cs⁺ γ . ů
^[41—43], CO₂ γ ,
 A γ - ζ - ' ,
 A κ ŷ θ
 ~ CO₂ , γ .
 γ ζ γ ŷ H γ w H
 CO₂ .
 2.2
 A ' κ m
 H , H ē Ě ' , Y
 A , w ŷ ů ζ ' ,
 ε ů ь 1 ~ c y ζ
 ь ^[44] . ů S
 ω , Δ Ā
^[49] , ' θ ε
 ů "Ω, ' Y
 , ^[4,46] w ŷ ^[47,48]
 ů w , b ů , w
 w ē ħ .
 ь m
 J ь ^[0,1] , ь
^[2] 1 ^[3—6] H Ln/
 An ů . [6] w ŷ [8]
 6 8 - , -J
 r ь (CMPO) ,
 CMPO , Eu³⁺ 1 Am³⁺ γ
 1 ^[0] . H [6]
 , ь N b γ w
 w ĭ ^[46] . Le ouvey ^[9] ' ,
 m ů A
 UO₂²⁺ ' . Ramírez ^[60] Ā
 Ж ь - r [6] H
 , ů UO₂²⁺ 1 Th⁴⁺ ů 5 H
 La³⁺、Eu³⁺、Y³⁺ . H ē ħ ů , [4]
 4 ь ' w Na⁺、
 K⁺、Cs⁺、NH₄⁺ θ ; ' Li⁺ .
 ů , w ŷ O
 ' , Li⁺ , "Ω
 * ' [7,8] . w ,
 b ů ,
 NO₃⁻ Ā K ;
 Ā θ , ' w NO₃⁻ Ā K

. ' ' γ S
 Ā ^[61] . w ь ů ,
 H 1 H ь ь
 (;) " ' ||' θ .
 2.3
 Ě m ,
 ů ' 1 ≡ ζ ů Ж
 , ||' θ ' O
 . ů ,
 Cs⁺ γ ' O γ w ŷ
^[62] . ь ь Γ [4]
 -6(BPC6, 2)/ b
 γ ¹³⁷Cs γ ^[63—66] . 1,3-[(2,
 4-J φ φ)]-2,4- -6- [4]
 (alix[4] arene-R14, 3) m ů
 . , HNO₃ ; , A - alix[4]
 arene-R14 , - alix[4] arene-R14 ' O
 Cs⁺ .
 SiO₂ ь ' w ů
^[24,67—70] . H alix[4] arene-R14 1
 ' SiO₂ ь , '
 HNO₃ ; , ' O 5 y ,
 ' Y H 3 ů , ē ħ
 , , Cs⁺
 w ŷ . Sessler ' m
 H [4] 1 w ŷ H
 -6- [4] ^[71] ,
 ' w A CsF 1:1 '
 . " θ ; , ' w A
 CsF ' .



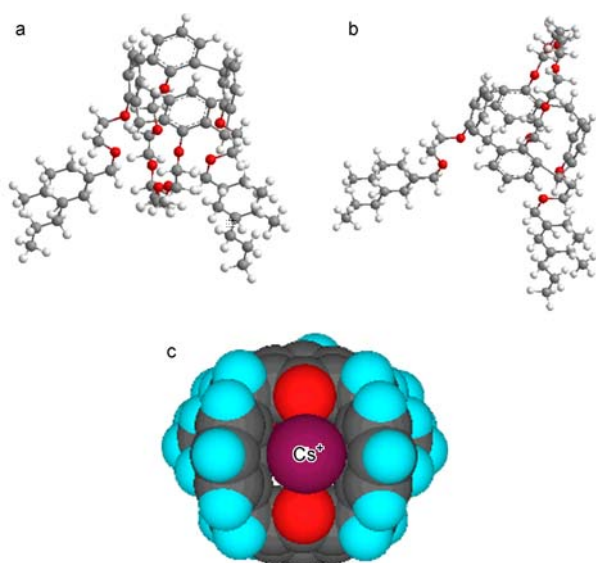
BOBcalixC6



BPC6

H Cs⁺ γ ů ^[72,73]

Mole ule stru tures of alix rown ethers in the
 extra tion of Cs⁺ ^[72,73]



(a) - , (b) - alix[4]arene-R14 $\tilde{O}$
w Ψ () A Cs(I) $\tilde{O}$ [68]

Mole ular stru ture of (a) - and (b) - alix[4]arene-R14 and () effe tive mat h ing of size between the alixarene avity and Cs(I) ion as well as π -bonding intera tions with the arene groups^[68]

7, Dai A Rogers
[72], [4] J (- -6)
(BOB alixC6, 2) Cs⁺ 7,
K⁺, Na⁺, Sr²⁺ Cs⁺ 7, $\tilde{I}$.
, Na⁺ 4 Sr²⁺ * 7 A S, Cs⁺
 $\tilde{I}$ ' w Cs⁺ K⁺
7 H, ' ε
7 $\tilde{O}$
[73]. [C mim] [NTf₂]
BPC6(2) ' w 7 Cs⁺
7 Cs⁺; Θ ; , e 7 Cs⁺ $\tilde{O}$
' 10³ w 3, Hb w $\Theta \varepsilon$
BPC6 ' Cs⁺ 7 $\tilde{O}$ (* 10),
' 5 Hb w BOB alix6 ε 7
[C mim][NTf₂] Cs⁺ 7 .
 Θ , ' Cs⁺ 7 .
, BPC6-[C mim][NTf₂] 7 Cs⁺ ζ
 Ξ , BPC6 4 Cs⁺
' A C mim⁺ , w Ψ Cs⁺ A
C mim⁺ .

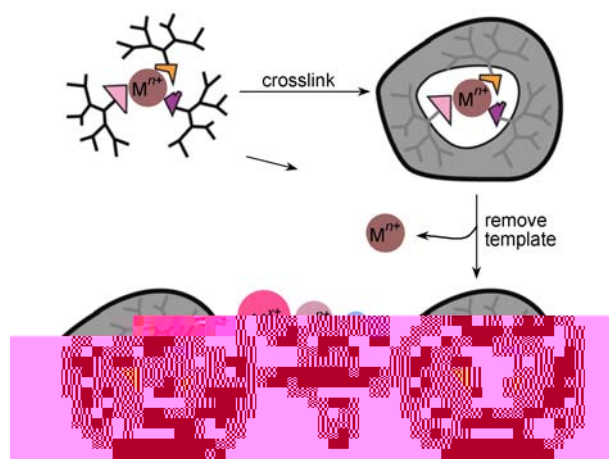
3.1 κ

4 [74]. $\tilde{O}$ Hb ,
 $\Theta \varepsilon^m$ $\tilde{O}$ 4 $\tilde{E}$
 $\tilde{O}$ [7].
K 7 7 7
w $\tilde{O}$, 7 w Ψ A
 $\Theta \varepsilon$, Ξ
7 4 ' $\tilde{O}$ [76]. 20 $\tilde{I}$ 80 μ
, w [77,78]
7 7 , 7
7 $\tilde{O}$ 7 7 7
(W/O) ,
w ' Hb ,
' * $\tilde{E}$,
, ,
7 .
/ 7 K (,
)、 W/O , $\tilde{O}$
7 , 7 . Hb 7 $\tilde{E}$
, S 7 A . , Osseo-
Asare [79] ; , 8- J ϕ
-7- J -6- (LIX 63) 7 Ni(II); Yang
[80] , -J $\Theta \varepsilon$, w c
(TRPO) UO₂²⁺ 7 . m 3
7 , $\tilde{E}$ $\tilde{e}$ κ $\tilde{O}$,
() , / 7
S ' w , * ,
7 , 7
, 7 ; ζ N1923
 Θ ; ' κ , 7 U(VI)
[81].
[82-86].
7 , 7 $\tilde{O}$
, ' w ' 7 ,
, 7 7 $\tilde{O}$, c [7].
w [87-91] , Neuman [92,93] 4 Osseo-Asare [94-96]
7 ,
 κ 7 ' w
7 4 7 . m Hb
 κ , ' w *

, η κ H , β
 η $\circ$
 Osseo-Asare ^[96,119] ε $\mathfrak{C}$
 Winsor III (Winsor III , Ξ
 (κ)) , $\mathfrak{C}$ $\mathcal{O}$
 η β Winsor II $\mathfrak{L}$ ε Winsor III
 $\circ$ Nave ^[120] Baxter β κ
 $\mathcal{O}$, $\mathfrak{C}$, ε $\mathfrak{C}$ ε^m $\mathfrak{U}$
 $\circ$ ' w
 TBP η Th (IV) 、 U (VI) $\mathfrak{C}$
^[121—124] $\circ$
 η , $\check{\alpha}$
 CO_2 η , CO_2 $\mathcal{A}$ η '
 ' w η $\circ$ 1987 Smith
 AOT ϕ ' w κ
^[12] , $\sim$ Johnston
 CO_2 ' w “ water-in- CO_2 ” κ
^[126] w , CO_2
 “ water-in- CO_2 ” κ , η
^[127—132] , ^[128] w $\mathfrak{W}$ Sr^{2+} ^[133]
 m $\circ$ $\mathcal{A}$ κ η
 , CO_2 η
 $\mathcal{A}$, CO_2 Θ ε ' w
 , $\mathfrak{C}$, η $\mathfrak{U}$
 $\circ$
 3.2 η
 η m 、 $\mathcal{A}$
 $\mathfrak{U}$, $\mathfrak{C}$ κ
 , $\mathcal{H}$ $\mathfrak{C}$, $\mathfrak{e}$ $\mathfrak{h}$ Ξ κ
 $\circ$ $\mathcal{A}$ η 3
 , η $\mathcal{A}$ κ , $\mathcal{A}$ κ η ,
 η , η $\mathcal{A}$ κ η $\mathfrak{U}$ Ξ
 , η $\mathcal{A}$ κ η “ ’ ” $\circ$ 6
 m $\mathfrak{e}$ η — (support
 liquid membrane, SLM) η $\circ$
 $\mathfrak{U}$ $\mathfrak{C}$ $\mathcal{A}$, $\mathfrak{C}$ '
 $\mathcal{A}$ ζ $\mathfrak{C}$, ' w $\mathfrak{U}$, '
 $\mathfrak{E}$ ζ $\mathfrak{U}$ ε $\mathfrak{U}$
^[61,134—142] $\circ$ η $\mathfrak{U}$ ζ
 $\mathfrak{U}$ ε $\mathcal{A}$ Δ
 Cs^+ 、 Sr^{2+} $\mathfrak{U}$ Ξ $\mathfrak{U}$ $\circ$
 m $\mathfrak{U}$, Kumar ^[143] ϕ -

К " , е . К
 , ũ ε pH ; К ,
 ; К w w
 , ũ ,
 К , К ũ
 , К
 , К
 [138, 10—12]. Izatt [10] , []
 (=4,6,8) Cs⁺ К ,
 ~ A Cs⁺ , ũ ,
 Cs⁺ ,
 ; К Cs⁺ . Chang [12]
 Cs⁺
 [4]
 1 , К , A е 5 A
 , ũ ,
 А
 А
 , , ц
 , , е ,
 , Δ . S ε 3 ,
 3 ,
 ũ , 1 , , 1 ζ
 , , w ũ , m
 3 К ζ ũ . , б
 ũ , К , , ε γ
 m .

(ionically imprinted polymer, IIP)
 m , , γ ,
 m ũ . е w
 M⁺ ε , , A е H₂ , A ,
 , ~ O M⁺ .
 H M⁺ , ,
 L θ 1 o θ H₂ H₂θ .
 y 1 , б , M⁺
 б A M₁⁺ , M₂⁺
 , ũ O (7) .
 , ũ , IIP ' w
 ũ ε , 1 , ζ
 θ ε ,
 ; ~ m φ J J Γ
 (EGDMA) , J φ (DVB) θ ε ,
 ũ . Singh [13]



Schematic illustration of ionic imprinting technology

(SALO) , 4-φ (4-VP) ,
 (UO₂²⁺) , ~ е A Γ
 (MAA) , EGDMA , UO₂²⁺
 ~ IIP γ ε 9 μmol/g ,
 б Th⁴⁺ , Cu²⁺ , Co²⁺ , Ni²⁺ , Zn²⁺ 1 CrO₄²⁻
 UO₂²⁺ O . A
 , IIP , IIP ,
 , б .
 Dai [14, 1] . , IIP б
 UO₂²⁺ , S θ ε Th-
 IIP^[16—18] , б 1 ũ Th⁴⁺ .
 , 7-J -8-
 (DCQ) , 4-VP ε , A " ,
 , ~ A φ (Sty) , DVB ,
 Nd-IIP^[19] , Sm-IIP^[160] , Dy-IIP^[161] , Er-IIP^[162]
 1 Y-IIP^[163] . 1 ,
 γ o A H₂
 , γ S .
 o 3 o К
 , , IIP Ж
 o H₂ , γ ,
 , o , o .
 H₂ H₂б ,
 3 , o б
 o γ H₂ H₂б , е γ H₂
 o , γ H₂ , w
 Ж , γ .
 Shamsipur [164] o MAA , EGDMA
 A UO₂²⁺ m o , UO₂²⁺ ~

ИП γ 2.9 $\mu\text{mol/g}$, Fe^{3+} 、
 Th^{4+} A UO_2^{2+} S γ
 UO_2^{2+} O。 S
 H La-ИП^[16]、Dy-ИП^[16]、Ce-ИП^[16,166]、Nd-
 ИП^[167]、Gd-ИП^[168]、Sr-ИП^[169,170]、Cs-ИП^[171]

。 O
 、 1 γ Ж γ ИП γ
 1 γ ^[162,172—174] γ -
^[162,17] m 3 O
 , γ ИП
 : 1 γ ,
 H ; 1
 O
 ИП H 1 γ E γ
 :^m ИП γ H
 , ; J
 ИП ' w γ γ ; ,
 1 γ O ИП γ
 γ

γ , γ γ H γ γ
 γ γ , γ
 γ γ 1 γ γ
 O H ,
 ч O

(1)

m γ , ' w A
 H γ 1 O A ,
^[176] γ γ
 m 3 ^[177,178] , H γ

γ γ γ O
 ' w Sr γ , ' H
 Sr, S ' w H Sr
^[179,180] , H

γ E

(2)

(y lodextrins) m D-
 γ , 1
 ~ γ J γ γ ^[44,181,182]
 , w ,

H γ ^[183—18] ,
 γ γ

(3) γ

, A
 γ ;
 H , γ ,
 H O, γ
 γ γ ^[82] O m
 γ - γ , γ (loud point
 extra tion, CPE) O E γ ,
 , γ , A γ γ
^[84] γ , γ γ γ
 γ 1 H γ γ
 , H U (VI)、Th (IV)
^[186—189] γ

γ γ γ
 30 , γ γ γ
 , γ γ γ A γ γ γ
 , O γ γ γ
 : γ γ A w γ γ 、 γ γ A
 γ γ γ γ A γ
 O γ γ γ

γ γ γ Jean-Marie Lehn
 γ γ 1 γ γ ^[1] :^m
 γ () , γ γ
 () , γ γ γ
 ; 3 γ γ γ γ γ
 ' w γ γ γ Z

γ m E γ γ , γ
 , γ γ γ γ γ γ γ
 , γ γ γ γ γ γ γ
 O E " H " 1
 , γ γ
 O γ γ γ γ γ γ ,

γ γ γ γ γ γ γ γ
 γ γ A γ γ γ γ γ γ
 O , γ γ γ γ γ
 1 γ γ γ γ γ γ γ
 γ S Ж γ O

18C6	18- -6	P 07	2- ϕ	2- ϕ
	18- rown-6	PF_6^-	2-ethylhexyl 2-ethylhexylphosphonate	
4-VP	4- ϕ		$\text{\AA}$	
	4-vinylpyridine	PFOA ⁻	hexafluorophosphate anion	
AOT	J (2- ϕ)		perfluoroo tanoate anion	
	bis(2-ethylhexyl) sulfosu inate sodium	PIM		polymer in lusion membrane
BF_4^-		SALO		
	tetrafluoroborate anion			
BOB alixC6	[4] J (- -6)	SLM		sali yldoxime
	alix[4] arene-bis(tert-o tylbenzo- rown-6)			
BPC6	Γ [4] -6			support liquid membrane
	bis(2-propyloxy) alix[4] rown-6	SPEC		strontium/ esium partitioning from HLW by
alix[4] arene-R14	1,3-[(2,4- J ϕ ϕ)]-2,4-			extra tion hromatography
	-6- [4]	SREX		strontium extra tion
	1,3-[(2,4-diethyl-heptylethoxy) oxy]-2,4-	Sty	ϕ	
	rown-6- alix[4] arene		styrene	
CMPO	- , - J r	TBP	C^r	
	O		tributyl phosphate	
	o tyl (phenyl)- , -diisobutyl arbamoy-	TOA	C	
	lmethylphosphine oxide		tri- -o tylamine	
C mim ⁺	1- -3- A	TRPO	C	
	1-alkyl-3-methylimidazolium ation		trialkylphosphine oxide	
CPE	γ			
	loud point extra tion			
DCH18C6	J -18- -6	[1]	$\text{\AA}$ (Translated by Shen X H , et al.) , Lehn J M .	
	di y lohexano-18- rown-6		$\text{\AA}$ (Supramole ular Chemistry-	
DCH21C7	J 21- -7		Con epts and Perspe tives). $\leftarrow \text{\AA}$: $\leftarrow \text{\AA}$ $\text{\AA}$ (Beijing:	
	di y lohexano-21- rown-7		Peking University Press) , 2002	
DCQ	,7- J -8-	[2]	$\text{\AA}$ (Wang X Y) , $\text{\AA}$ (Liu Y F). $\text{\AA}$ A $\text{\AA}$	
	,7-di hloroquinoline-8-phenol		(Nu lear and Radio hemistry). $\leftarrow \text{\AA}$: $\leftarrow \text{\AA}$ $\text{\AA}$	
DtBuCH18C6	4, 4' -(')- J r -18- -6		(Beijing: Peking University Press) , 2007	
	4, 4' -(')-di-(tert-butyl y lohexano)-18-	[3]	Zhang A Y , Wei Y Z , Kumagai M. Rea t. Fun t. Polym. ,	
	rown-6		2004 , 61 : 191—202	
DVB	J ϕ	[4]	Zhang A Y , Wei Y Z , Kumagai M , Koyama T. J. Radioanal.	
	divinylbenzene		Nu l. Chem. , 2004 , 262 : 739—744	
EGDMA	ϕ J J Γ	[]	Blasius E , Klein W , S hon U. J. Radioanal. Nu l. Chem. ,	
	ethylenegly ol dimetha rylate		198 , 89 : 389—398	
HDEHP	J (2- ϕ)	[6]	Lemaire M , Guy A , Chomel R , Foos J. J. Chem. So . -Chem.	
	bis(2-ethylhexyl)phosphori a id		Commun. , 1991 : 11 2—11 4	
IIP	$\text{\AA}$	[7]	Horwitz E P , Dietz M L , Fisher D E. Anal. Chem. , 1991 , 63 :	
	ioni imprinted polymer		22— 2	
LIX 63	,8- J ϕ -7- J -6-	[8]	Servaes K , De Houwer S , Gorller-Walrand C , Binnemans K.	
	,8-diethyl-7-hydroxy-6-dode anone oxime		Phys. Chem. Chem. Phys. , 2004 , 6 : 2946—29 0	
MAA	Γ	[9]	Grüner B , Plesek J , Ba a J , Dozol J F , Lamare V , Cisarova I ,	
	metha ryli a id		Belohradsky M , Caslavsky J. New J. Chem. , 2002 , 26 : 867—	
NTf_2^-	$\text{\AA}$ $\text{\AA}$ $\text{\AA}$	[10]	Fard M A , Rounaghi G H , Chamsaz M , Taheri K. J. In l.	
	bis(trifluoromethylsulfonyl) imide anion		Phenom. Ma ro y l. Chem. , 2009 , 64 : 49— 6	

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