[Article]

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Uranium; Ionic liquid; Extraction third phase; Extraction selectivity; Extraction mechanism



 $\begin{array}{ccc} UO_2(NO_3)_2(CMPO)_2, & , \\ UO_2(NO_3)(CMPO)^+ & . \ Cocalia & ^{21} \\ (2,4,4- &) & (2- &) \end{array}$

 $\begin{array}{ccccccc} (\text{HDEHP}) & C_{10}\text{mimNTf}_2 & UO_2^{2+} & Am^{3+} \\ & , & UO_2^{2+} & , \\ Am^{3+} & . & Shen^{22} & Rout^{23} \\ & & & C_n\text{mimNTf}_2 & C_n\text{mimPF}_6 \\ UO_2^{2+} & Th^{4+} & Pu^{4+} & Am^{3+} & , \\ & & & & . \\ (\text{TBPO}) & & , \end{array}$

 $UO_2(NO_3)_2$. , UO_2^{2+} ,

C_nmimNTf₂ 1- -3- (C_nmimBr) (LiNTf₂) .¹ C_nmimBr LiNTf₂ , 99%. TOPO Aldrich , 99%. TBPO , 98%. UO₂(NO₃)₂· 6H₂O Chemapol, 98.5%. . Sr(NO₃)₂, CsNO₃, ZrOCl₂· 8H₂O, Cr(NO₃)₃.

U	O_2^{2+}			-
(Ie	CP-AES, Le	eeman, US	A)	
NO_3^-	Dionex 1	model ICS	5- 900	(Dionex
Corporation, USA	.)	,		Chrome-

	1 U	JO_{2}^{2+}		
(SLW) ¹¹ Table 1 Compositions of simulated liquid waste (SLW) used in the selective extraction experiment of UO ²⁺¹¹				
Element	$C/(g \cdot dm^{-3})$	Element	C/(g· dm ⁻³)	
Sr	0.54	Ru	0.38	
Cs	0.54	Ni	0.17	
Zr	0.68	Nd	1.00	
Cr	0.34	U	2.38	
Re	0.15			

No.5	:	-	$UO_2(NO_3)_2$
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leon 7.0 . U-3010 (HITA-CHI, Japan) - . (CEM, USA)



 $\begin{array}{cccc} 1 & TBPO & C_{\pi}mimNTf_{2} \ (n=2, \, 4, \, 6, \, 8) & 10 \ mmol\cdot \ L^{-1} \\ & UO_{2}(NO_{3})_{2} & C_{4}mimNTf_{2} & 10 \ mmol\cdot \ L^{-1} \\ & UO_{2}(ClO_{4})_{2} & (E) \end{array}$

Fig.1 Dependence of the extraction efficiency (E) of 10 mmol· L⁻¹ UO₂(NO₃)₂ in C_nmimNTf₂ (n=2, 4, 6, 8) and 10 mmol· L⁻¹ UO₂(ClO₄)₂ in C₄mimNTf₂ on the concentration of TBPO

TBPO (C_nmimNTf₂) $UO_2(NO_3)_2$ TBPO $C_n mimNTf_2$, . $UO_2(NO_3)_2$ C_nmimNTf₂ NO_3^- TBPO $UO_2(NO_3)_2$ C_nmimNTf₂ . TOPO $C_n mimNTf_2$ $UO_2(NO_3)_2$ C_nmimNTf₂). (

2 TBPO C₄mimNTf₂ $UO_2(NO_3)_2$ 2 . HNO₃ $0.1 \text{ mol} \cdot L^{\scriptscriptstyle -1}$, 85% HNO₃ $0.1 \text{ mol} \cdot L^{-1}$, HNO_3 $UO_2(NO_3)_2$ HNO₃ 33 , HNO₃

TBPO , TBPO , 2 , $LiNTf_2$. TBPO $C_4mimNTf_2$ $UO_2(NO_3)_2$



 $\label{eq:Fig.2} \begin{array}{ll} Dependence \ of \ the \ extraction \ efficiency \ of \ 10 \ mmol} \\ L^{-1} \ UO_2(NO_3)_2 \ by \ TBPO-C_4 mimNTf_2 \ on \ the \ concentration \ of \ HNO_3 \ and \ salts \ in \ the \ aqueous \ phase \end{array}$

).

(

 UO_{2}^{2+} $TBPO-C_4mimNTf_2$ U Cs Re Zr Nd Sr Cr Ru Ni 9 , Sr Cs , Re Tc Nd . $TBPO-C_4mimNTf_2$ U Cs Re Zr 0-5 mol· L⁻¹ 3 $1 \text{ mol} \cdot L^{-1}$ U, Re Cs Zr U TBPO-C₄mimNTf₂ U 1 mol· L^{-1} U HNO₃ TBPO-C₄mimNTf₂ U Nd (4), Zr Nd U

.

3.5.1



:

C₄mimNTf₂









	,		
C"mimľ	NTf_2 (<i>n</i> =2,	4, 6, 8) NO	3
	5	$UO_2(NO_3)_2$	(pH=3.3),
	,		NO_3^-
,	TBPO	0	100 mmol· L^{-1}
		NO_3^-	
		. C ₂ mim	NTf_2 ,
NO_3^-		2-	$-3 \text{ mmol} \cdot L^{-1}$,
		TBPO	$C_2 mimNTf_2$
UO ₂ (NO	$(D_3)_2$	NO_3^-	;

 NO_3^-

 NO_3^-

,



Fig.5 Relationship between the concentrations of NO₃⁻ and TBPO in C_nmimNTf₂ (n=2, 4, 6, 8) phase after TBPO extracting 10 mmol· L⁻¹ UO₂(NO₃)₂





6 TBPO $C_n \text{mimNTf}_2$ (*n*=2, 4, 6, 8) 10 mmol· L⁻¹ UO₂(NO₃)₂ NO₃⁻ UO₂²⁺

Fig.6 Linear relationship between the concentrations of NO₃⁻ and UO₂²⁺ in C_nmimNTf₂ (*n*=2, 4, 6, 8) phase after TBPO extracting 10 mmol· L⁻¹ UO₂(NO₃)₂

 $UO_2(TBPO)_n(NO_3)^+$. $C_2 mimNTf_2$, $C_4 mimNTf_2$ $C_2 mimNTf_2$ $C_6 mimNTf_2$. , 1/4, NO_3^- 1/40.55 0.72, 1 $C_n mimNTf_2$ $C_8 mimNTf_2$ $NO_3^ UO_2(TBPO)_n(NO_3)^+$ $UO_2(TBPO)_n^{2+}$, 437.4 3.0 (a) 2.0 10⁻⁸ Intensity 1.5 1.0 502.1 986.6 0.5 145.8 2<u>18</u> 768.4 768.4 1204.5 0.0 200 400 600 800 1000 1200 1400 1600 1800 m/z 2.0 437.4 (b) 10⁻⁸ Intensity 1.5 558.2 1.0 986.6 0.5 1204.5 768.4 846.4 5.8 279.1 0.0 200 600 800 1000 1200 1400 1600 1800 400 m/z 986.6 2.0 (C) 10⁻⁸ Intensity 1.5 1.0 437.4 0.5 1204.5 1101.7 328.9 614.2 768.4 145.8 0.0 200 800 1600 1800 400 600 1000 1400 1200 mår: ļ iirn 1⁰1101 7 TBPO (200 mmol· L^{-1})- C_n mimNTf₂ (n=2 (a), n=4 (b), n=6 (c), n=8 (d)) 40 mmol· L⁻¹ UO₂(NO₃)₂ (ESI-MS) TBPO (200 mmol· L⁻¹)-C₂mimNTf₂ 40 mmol· L^{-1} $UO_2(NO_3)_2$ ESI-MS

Fig.7 Positive ion spectra of electrospray ionization mass spectra (ESI-MS) of IL phases after extracting 40 mmol· L⁻¹ $UO_2(NO_3)_2$ by TBPO (200 mmol· L⁻¹)-C_nmimNTf₂ (n=2 (a), n=4 (b), n=6 (c), n=8 (d)) and negative ion spectrum of ESI-MS in the case of C₂mimNTf₂ (e)

(e)

No.5

3.5.2 TBPO	3 200 mmol· L ⁻¹ TBPO-C _n mimNTf ₂ (<i>n</i> =2, 4, 6, 8) 40 mmol· L ⁻¹ UO ₂ (NO ₃) ₂ ESI-MS
UO ₂ (NO ₃) ₂ , (ESI-MS) . 7 TBPO UO ₂ (NO ₃) ₂ ESI-MS .	Table 3Peak positions associated in the ESI-MS of IIphase after extracting 40 mmol· L^{-1} UO2(NO3)2 by 200mmol· L^{-1} TBPO-C _n mimNTf2 (n=2, 4, 6, 8)
7(a) $C_2 \text{mim} NT f_2$ $UO_2 (TBPO)_2 (NO_2)^+$	IL Species <i>m/z</i>
$[IIO (TPDO)^{2+} NTf^{-1+} (m/7)]$	$C_2 mimNTf_2$ [UO ₂ (TBPO) ²⁺ ₃ +NTf ₂] ⁺ 1204.5
$[002(1010)_3 + 1011_2]$, (<i>mu2</i>)	$UO_2(TBPO)_3(NO_3)^+$ 986.6
986.6 1204.5, TBPO-	$UO_2(TBPO)_2(NO_3)^+$ 768.4
$C_2 mimNTf_2$ $UO_2(NO_3)_2$	$[2C_2 mim^+ + NTf_2^-]^+$ 502.1
. (<i>m</i> /	[2TBPO+H] ⁺ 437.4
z=437.4) [2TBPO+	NTf ₂ 279.9
H ¹⁺ TBPO-C ₄ mimNTf ₂ TBPO-C ₆ mimNTf ₂	$C_4 \text{mimNTf}_2$ [UO ₂ (TBPO) ²⁺ +NTf ² ₂] ⁺ 1204.5
$IIO(NO) \qquad (7(h c))$	$UO_2(1BPO)_3(NO_3)^{+}$ 986.6
$UO_2(INO_3)_2$ (7(0, c)),	$UO_2(1BPO)_2(1NO_3)$ 708.4 [2C mim ⁺ NTf ⁻] ⁺ 558.2
$UO_2(1BPO)_3(NO_3)^{-}(m/z=986.6) [UO_2(1BPO)_3^{-}+$	$[2C_4 \text{IIIIII} + N \Pi_2]$ 556.2 $[2TBPO+H]^+$ 437.4
NTf_2] ⁺ (<i>m</i> / <i>z</i> =1204.5) , TBPO-	NTf ₂ 279.9
C ₈ mimNTf ₂ $UO_2(TBPO)_3(NO_3)^+ (m/z =$	$C_6 mimNTf_2$ [UO ₂ (TBPO) ₃ ²⁺ +NTf ₂] ⁺ 1204.5
986.6) . $TBPO-C_8mimNTf_2$	$UO_2(TBPO)_3(NO_3)^+$ 986.6
$UO_{2}(TBPO)_{3}(NO_{3})^{+}$.	$UO_2(TBPO)_2(NO_3)^+$ 768.4
C mimNTf ₂ $(n-2, 4, 6, 8)$ U(Ω_{1} (TBPO) ₂ (NO ₂) ⁺	$[2C_6 mim^+ + NTf_2^-]^+$ 614.2
(m/r - 0.06.6)	[2TBPO+H] ⁺ 437.4
(<i>m</i> /2=980.0) ,	NTf ₂ 279.9
, $C_8 m m N T t_2$ (7(d))	$C_8 mimNTf_2^*$ UO ₂ (TBPO) ₃ (NO ₃) ⁺ 986.6
, .	$UO_2(1BPO)_2(NO_3)^*$ 768.4
$UO_2(TBPO)_3(NO_3)^+$	$\frac{1}{1} \frac{1}{12} $
. $C_n \text{mimNTf}_2$ (<i>n</i> =2, 4, 6, 8)	The peak intensity of $UO_2(TBPO)_3(NO_3)$ in C_8 mimin Π_2 extraction system is very high which results in the disappear of peak of
m/z=279.9	$[2C_smim^++NTf_2^-]^+$.
$NTf^{-}(7/e) C mimNTf$	
$\frac{1}{112} (7(e) C_2 \frac{1}{1111} \frac{1}{12}$	2 LiNTf.
). 5.	
3.5.3 -	, C_4 mimiN I I_2
-	((2)), NTf_2^-
(8),	$C_4 mimNTf_2$,
, 422 nm (C_2 mimNTf ₂)	$C_4 \min^+$, ((3) (4)
$425 \text{ nm} (C_{\circ} \text{mim} \text{NTf}_{\circ})$	¹² HNO
TPDO UO(NO)	, i into
	, LINO ₃
. NO ₃ ,	$, NO_3^-$
NO_3^-	
,	
	$C_{*}mim^{+}$
$C \min NTf_{L_{a}} C \min^{+} + NTf $ (2)	€ _n iiiiii
$U_n^{2+} + N_n + 2TDDO + C + +$	
$UO_{2aq}^{-1} + NO_{3aq} + 31BPO_{IL} + C_n mim_{IL}^{-1} $ (3)	,
$UO_2(TBPO)_3(NO_3)^+_{IL} + C_n \min_{IL}^+$, $C_8 mimNTf_2$

 $UO_{2aq}^{2+} + 3TBPO_{IL} + 2C_nmim_{IL}^+$

 $UO_2(TBPO)_{3 IL}^{2+} + 2C_n mim_{aq}^{+}$ (4)

6, 8)

of IL 200

204.5 986.6 768.4 614.2 437.4 279.9 986.6 768.4 279.9 raction k of

JTf_2^- , 4)) INO₃ , ,

 Tf_2 (3) , $UO_2(TBPO)_3(NO_3)^+$ $C_8 mimNTf_2$



8 TBPO (40 mmol· L⁻¹)-C_nmimNTf₂ (n=2, 4, 6, 8) 10 mmol· L⁻¹ UO₂(NO₃)₂ -

Fig.8 UV-Vis absorption spectra of IL phase after extracting 10 mmol: L^{-1} UO₂(NO₃)₂ by TBPO (40 mmol: L^{-1})-C_nmimNTf₂ (n=2, 4, 6, 8)

C₂mimNTf₂ 2 $UO_2(TBPO)_3^{2+}$ (4)C_nmim⁺ . (3) $NO_3^- UO_2^{2+}$ C₂mimNTf₂ C₂mimNTf₂ (4) (3) $(UO_2(TBPO)_3(NO_3)^+)$ $UO_2(TBPO)_3^{2+})$ C₈mimNTf₂ (3) UO_2 $(TBPO)_{3}(NO_{3})^{+}$.

 $C_4 mimNTf_2$

U Nd, Zr

. , NO₃

•

. , NO_3^- . $NO_3^ C_8mimNTf_2$

4, 6) $UO_2(TBPO)_3(NO_3)^+$, $C_n minNTf_2$ (*n*=2, $UO_2(TBPO)_3(NO_3)^+$ $UO_2(TBPO)_3^{2+}$

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