

Macrolide Synthesis through Intramolecular Oxidative Cross-Coupling of Alkenes

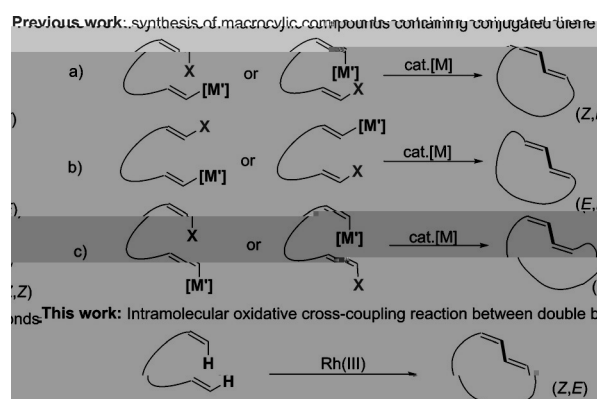
Bing Jiang, Meng Zhao, Shu-Sen Li, Yun-He Xu,* and Teck-Peng Loh*

Abstract: A Rh^{III}-catalyzed intramolecular oxidative cross-coupling between double bonds for the synthesis of macrolides is described. Under the optimized reaction conditions, macrocycles containing a diene moiety can be formed in reasonable yields and with excellent chemo- and stereoselectivity. This method provides an efficient approach to synthesize macrocyclic compounds containing a 1,3-conjugated diene structure.

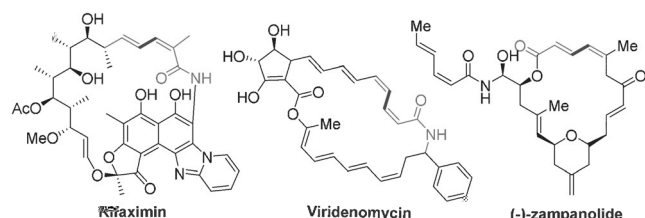
Macrocycles are an important class of compounds since they feature widely in many natural products and pharmaceuticals.^[1] Common strategies utilized to construct these macrocyclic compounds are macrolactonizations,^[2] macroaldolizations,^[3] and macrolactamizations.^[4] In recent decades, transition-metal-catalyzed cross-coupling reactions to forge new C–C or C–X (X = heteroatom) bonds for the construction of large rings have also emerged as powerful tools.^[5] In recent years, the strategy of ring-closing metathesis (RCM) has also become a popular way to construct large rings containing double bonds, albeit with difficulty in controlling the E/Z stereoselectivity.^[6] There are also numerous macrocyclic natural products, such as rifaximin, viridenomycin, and (–)-zampanolide, that contain a conjugated diene moiety.^[7]

However, methods to construct this type of macrocycles are rare. In addition to the traditional strategies involving

step-by-step synthesis of the conjugated double bonds, a few examples involving the synthesis of such compounds through transition-metal-catalyzed coupling reactions using organometallic reagents to connect different double bonds have also recently been reported (Scheme 1a and b).^[8] Inspired by the



Scheme 1. Synthesis of macrocyclic compounds containing a 1,3-conjugated diene moiety.



[*] B. Jiang, M. Zhao, S.-S. Li, Prof. Dr. Y. H. Xu, Prof. Dr. T. P. Loh
Hefei National Laboratory for Physical Sciences at the Microscale
iChEM (Collaborative Innovation Center of Chemistry for Energy
Materials), University of Science and Technology of China
Hefei, Anhui 230026 (China)
E-mail: xyh0709@ustc.edu.cn
teckpeng@ustc.edu.cn

Prof. Dr. T. P. Loh

Institute of Advanced Synthesis, Jiangsu National Synergetic
Innovation Center for Advanced Materials, Nanjing Tech University
Nanjing, Jiangsu, 210009 (China)
and

Division of Chemistry and Biological Chemistry, School of Physical
and Mathematical Sciences, Nanyang Technological University
Singapore 637616 (Singapore)

recent advances on direct transition-metal-catalyzed C–H bond functionalization,^[9] especially the pioneering work reported by Glorius and co-workers on Rh-catalyzed oxidative cross-coupling reactions between electron-deficient alkenes,^[9a,b] and as part of our long-standing interest in the alkene-alkene coupling reactions,^[10] we envisage that an intramolecular cross-coupling between two different double bonds will provide an efficient method to access this class of compounds. Herein, we report a Rh^{III}-catalyzed macrocyclization reaction through intramolecular oxidative cross-coupling between double bonds. Macrocyclic products with large ring sizes can be obtained in reasonable yields through this strategy. This method is atom-economical and the products obtained can be converted into other useful functional compounds. In addition, this is the first example of intramolecular oxidative cross-coupling between double bonds.

To evaluate the feasibility of the rhodium-catalyzed intramolecular cross-coupling reaction between alkenes, we chose the easily available 6-methacrylamidoethyl acrylate (11) for the model reaction. Optimization of the reaction conditions was carried out by employing [RhCp*Cl₂]₂ as the catalyst, and the results were summarized in Table 1. Pleasingly, the 14-membered-ring product 21 could be obtained in 8 % yield and with a single Z,E configuration in the presence of 5 mol % [RhCp*Cl₂]₂ catalyst and Cu(OAc)₂·H₂O (2.0 equiv) as an oxidant in 1,2-dichloroethane (DCE) (0.01 M) at 100 °C upon stirring for 24 h under argon atmos-

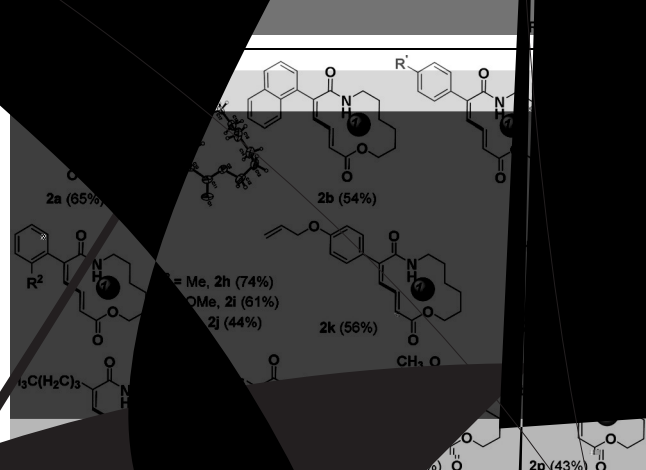
Entry	Additive	Solvent	Conc. [mol L ⁻¹]	T [°C]	Yield ^[b] [%]
1	–	DCE	0.01	100	8
2	–	Toluene	0.01	100	19
3	–	THF	0.01	100	26
4	–	Acetone	0.01	100	42
5 ^[c]	–	Acetone	0.01	100	38
6	–	Acetone	0.01	80	34
7	–	Acetone	0.01	120	29
8	–	Acetone	0.20	100	25
9	NaOAc (1.5 equiv)	Acetone	0.01	100	32
10	HOAc (15 equiv)	Acetone	0.01	100	39
11	AgPF ₆ (20 mol %)	Acetone	0.01	100	21
12	AgSbF ₆ (20 mol %)	Acetone	0.01	100	11
13 ^[d]	–	Acetone	0.01	100	28
14 ^[e]	–	Acetone	0.01	100	46
15 ^[f]	–	Acetone	0.01	100	24
16 ^[g]	–	Acetone	0.01	100	42
17 ^[e,h]	–	Acetone	0.01	100	nr
18 ^[e]	NaBARF (20 mol %)	Acetone	0.01	100	52
19 ^[e]	NaBARF (40 mol %)	Acetone	0.01	100	55
20 ^[e,i]	NaBARF (40 mol %)	Acetone	0.01	100	65 (69)

[a] Reaction was run under the following reaction conditions: 0.15 mmol **1**, 5 mol % $[\text{RhCp}^*\text{Cl}_2]_2$, 0.375 mmol $\text{Cu}(\text{OAc})_2 \cdot \text{H}_2\text{O}$ and 0.06 mmol NaBARF in 15 mL acetone at 100 °C for 24 h under air atmosphere. Yields in parentheses are yields of isolated product. NaBARF is sodium tetrakis[3,5-bis(trifluoromethyl)phenyl]borate. [b] Yield was determined by ^1H NMR using mesitylene as an internal standard. [c] 2.5 mol % $[\text{RhCp}^*\text{Cl}_2]_2$ (2.0 equiv) was used. [d] 2.5 mol % $[\text{RhCp}^*\text{Cl}_2]_2$ (2.0 equiv) was used. [e] Under oxygen atmosphere. [f] Under oxygen atmosphere. [g] 5 mol % $\text{Rh}(\text{CH}_3\text{CN})_3\text{Cp}^*(\text{PF}_6)_2$ (2.5 equiv) was used.

phenylboronic acid (1.0 equiv), Cu(OAc)₂·H₂O (1.0 equiv), NaBARF (1.0 equiv), and Na₂CO₃ (1.0 equiv) in CH₂Cl₂ (5 mL) at 100 °C (Scheme 1). The reaction mixture was cooled to room temperature and extracted with CH₂Cl₂ (3 × 5 mL). The combined organic layers were dried over anhydrous Na₂SO₄ and concentrated under reduced pressure. The residue was purified by silica gel chromatography (40% EtOAc in CH₂Cl₂) to give the desired product 21 in 52% yield (entry 18). Finally, the desired product 21 was isolated in 21% yield upon increasing the loading of NaBARF (4.0 equiv) and Cu(OAc)₂·H₂O (2.5 equivalents; Table 1, entries 19–20). Control experiments indicated that all

After determining the mechanism of the reaction, a further investigation of the scope and limitations of this intramolecular cross-coupling was undertaken. First, a series of reactions to examine the effect of substituents on the intramolecular cross-coupling was undertaken (Table 1). The first part (Table 1a) shows the effect of substituents on the intramolecular cross-coupling of 1a. The second part (Table 1b) shows the effect of substituents on the intramolecular cross-coupling of 1b. The third part (Table 1c) shows the effect of substituents on the intramolecular cross-coupling of 1c. The fourth part (Table 1d) shows the effect of substituents on the intramolecular cross-coupling of 1d. The fifth part (Table 1e) shows the effect of substituents on the intramolecular cross-coupling of 1e. The sixth part (Table 1f) shows the effect of substituents on the intramolecular cross-coupling of 1f. The seventh part (Table 1g) shows the effect of substituents on the intramolecular cross-coupling of 1g. The eighth part (Table 1h) shows the effect of substituents on the intramolecular cross-coupling of 1h. The ninth part (Table 1i) shows the effect of substituents on the intramolecular cross-coupling of 1i. The tenth part (Table 1j) shows the effect of substituents on the intramolecular cross-coupling of 1j. The eleventh part (Table 1k) shows the effect of substituents on the intramolecular cross-coupling of 1k. The twelfth part (Table 1l) shows the effect of substituents on the intramolecular cross-coupling of 1l. The thirteenth part (Table 1m) shows the effect of substituents on the intramolecular cross-coupling of 1m. The fourteenth part (Table 1n) shows the effect of substituents on the intramolecular cross-coupling of 1n. The fifteenth part (Table 1o) shows the effect of substituents on the intramolecular cross-coupling of 1o. The sixteenth part (Table 1p) shows the effect of substituents on the intramolecular cross-coupling of 1p. The seventeenth part (Table 1q) shows the effect of substituents on the intramolecular cross-coupling of 1q. The eighteenth part (Table 1r) shows the effect of substituents on the intramolecular cross-coupling of 1r. The nineteenth part (Table 1s) shows the effect of substituents on the intramolecular cross-coupling of 1s. The twentieth part (Table 1t) shows the effect of substituents on the intramolecular cross-coupling of 1t. The twenty-first part (Table 1u) shows the effect of substituents on the intramolecular cross-coupling of 1u. The twenty-second part (Table 1v) shows the effect of substituents on the intramolecular cross-coupling of 1v. The twenty-third part (Table 1w) shows the effect of substituents on the intramolecular cross-coupling of 1w. The twenty-fourth part (Table 1x) shows the effect of substituents on the intramolecular cross-coupling of 1x. The twenty-fifth part (Table 1y) shows the effect of substituents on the intramolecular cross-coupling of 1y. The twenty-sixth part (Table 1z) shows the effect of substituents on the intramolecular cross-coupling of 1z.

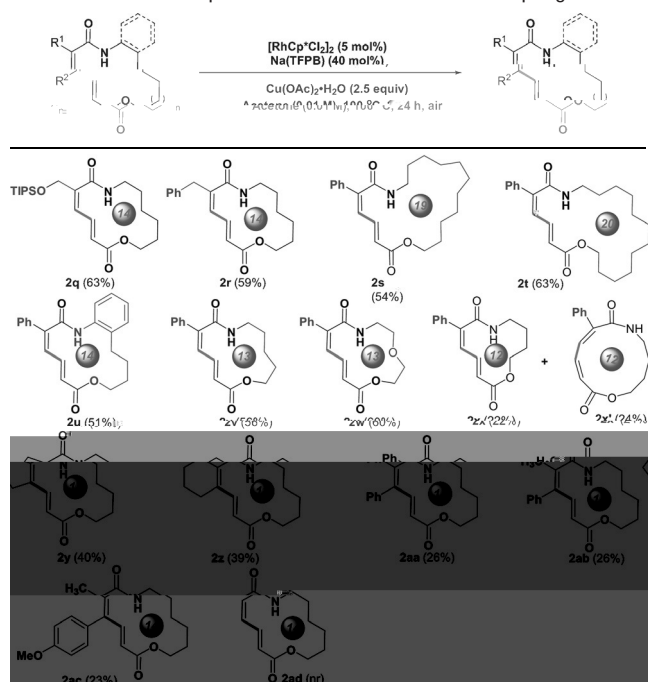
R^1



0.1% [RhCp*Cl₂], 0.375 mmol
in 15 mL acetone at 100°C for
parentheses are yields of isolated

desired 14-membered macrocycles were obtained in moderate to high yields from acrylate and acrylamide substrates with an aliphatic substituent at the a-position. Both acrylate and acrylamide substrates containing withdrawing functional groups such as cyano, ester, and amide, and a phenyl ring. The corresponding macrocyclic products were obtained in moderate to good yields (74% yields). The structure of 2x was confirmed by X-ray crystallography and single-crystal X-ray diffraction analysis.^[16] Furthermore, substrates with aliphatic substituents at the a-position of acrylate and acrylamide were also tested (2l–2o). It was found that substrates with a methyl, ethyl, propyl, isopropyl, n-butyl, ethoxymethyl, (triisopropylsilyl)oxy, and benzyl group all reacted well and furnished the corresponding macrocyclic products in moderate to good yields (2p–2t). The current annulation method allowed to access even bigger macrocyclic rings upon tethering a longer or shorter chain between the acrylate and acrylamide fragments (2s–2x/2x' and 2y–2z). It is worth noting that apart from the formation of product 2x in 22% yield with a Z,E configuration, another 14-membered macrocyclic product 2x' with a Z,Z configuration was also isolated in 24% yield when the acrylamide fragment 1x was applied in this reaction. Moreover, in contrast to the previous result,^[10b] substrate 1u, which has an N-phenyl acrylamide moiety, also showed good reactivity, and the desired 14-membered ring product 2u was obtained in 51% yield. Further studies showed that this method was not limited to a-

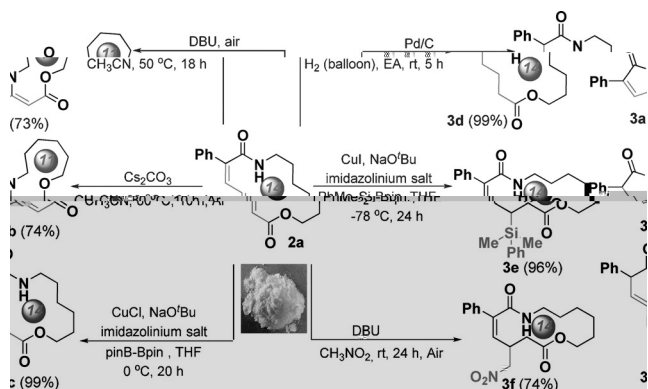
Table 3: Substrate scope for the intramolecular cross-coupling reaction



[a] Reaction was run under the following reaction conditions: 0.15 mmol **1**, 5 mol % $[\text{RhCp}^*\text{Cl}_2]_2$, 0.375 mmol $\text{Cu}(\text{OAc})_2 \cdot \text{H}_2\text{O}$ and 0.06 mmol NaBARF in 15 mL acetone at 100 °C for 24 h under air atmosphere. Yields in parentheses are yields of isolated product.

substituted substrates, but was also compatible with a,b-disubstituted acrylamide derivatives. However, these substrates only exhibited low reactivity along with low conversion. Acrylamides with a cyclopentenyl and a cyclohexenyl moiety afforded the desired bicyclic compounds in 40 % and 39 % yields, respectively (**2y–2z**). Low yields of the products **2aa** and **2ab** were observed when a-phenyl-b-phenyl- and a-methyl-b-aryl-substituted acrylamides were subjected to this reaction. Unfortunately, the unsubstituted substrate **1ad** did not give any desired product under the optimized reaction conditions, even with a prolonged the reaction time.

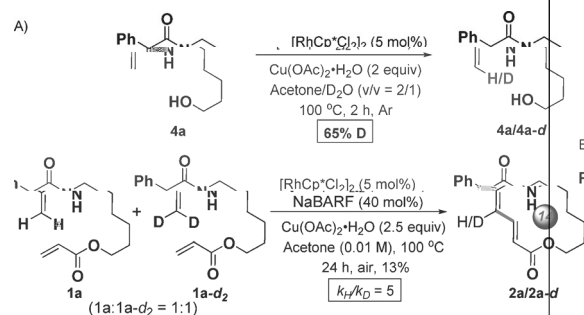
To explore the synthetic utility of this method, derivatization reactions of macrocycle **2a** were investigated (Scheme 2). Treatment of **2a** with 1,8-diazabicyclo[5.4.0]undec-7-ene



Scheme 2. Derivatization of the cross-coupling macrocycle product.

(DBU) in acetonitrile afforded the bicyclic compound **3a** in 73 % yield through intramolecular Michael addition and dehydrogenation processes.^[11] In contrast to this result, if **2a** was treated with Cs₂CO₃, the more stable Michael addition product **3b** could be obtained in 74 % yield.^[12] In addition, the 1,4-hydrosilylation of acrylate fragment product **3e** was obtained in 96 % yield under the NHC-Cu-O^tBu catalytic system.^[13] A similar chemoselective 1,4-Michael addition reaction happened when nitromethane was used as a nucleophile under basic conditions,^[14] and the desired product **3f** was isolated in 74 % yield. In addition, a highly stereoselective reductive product **3c** with E configuration was obtained in excellent yield when using a bis(pinacolato)diboron reagent and a copper catalyst. This result is attributed to a 1,6-conjugated boryl addition and a sequential allylic boronate reduction. Furthermore, hydrogenation reaction of **2a** catalyzed by Pd/C led to the product **3d** in a high yield.

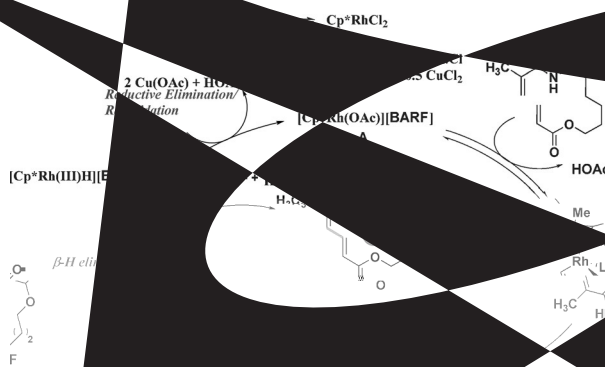
Finally, to gain some preliminary understanding of the mechanism, an isotope-labeling experiment with acrylamide **4a** was carried out. A remarkable Z-selective olefinic H/D exchange was detected with addition of D₂O, which implies a reversible cyclometalation process [Scheme 3 A]. Furthermore, a kinetic isotope effect (KIE) value ($k_{\text{H}}/k_{\text{D}} = 5$) was observed in the competitive cross-coupling reactions of **1a** and **1a-d₂**, thus suggesting that the alkenyl C–H bond activation might be rate-determining step [Scheme 3 B].



Scheme 3. Isotope-labeling experiments for KIE studies.

Therefore, a plausible mechanism to account for the Rh^{III}-catalyzed macrocyclization was proposed on the basis of our experimental results and literature precedent (Scheme 4).^[15] Anion exchange of $[\text{RhCp}^*\text{Cl}_2]_2$ with NaBARF and $\text{Cu}(\text{OAc})_2 \cdot \text{H}_2\text{O}$ afforded the reactive species A, which selectively activates the Z-olefinic C–H bond of acrylamide derivative **1** to form an vinylrhodium(III) species B. Subsequent intramolecular coordination and migratory insertion of acrylate form intermediates C and then D, which undergoes b-hydride elimination to provide the macrocycle with concomitantly release a Rh^{III} hydride species. Reductive elimination followed by reoxidation of $[\text{Rh}^I]$ to $[\text{Rh}^{III}]$ by $\text{Cu}(\text{OAc})_2 \cdot \text{H}_2\text{O}$ would release the active catalyst for the next catalytic cycle.

In summary, we have accomplished the first intramolecular Rh^{III}-catalyzed oxidative cross-coupling reaction between double bonds. This work provides a rapid and atom-economical synthetic pathway to form macrolide compounds with a Z,E-configured diene moiety. This work also



Scheme 1. Mechanism.

features, chemoselectivity and stereoselectivity, good functional group compatibility, and versatile utility of the diene fragment in further derivation reactions. In-depth application studies of this intramolecular oxidative cross-coupling reaction in organic synthesis are underway in our laboratory.

Acknowledgements

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interest

no conflict of interest.

Keywords: cross-coupling; homogeneous catalysis; macrocyclic; palladium.

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